

## **$\mu$ -(Furan-2-carbaldehyde azine)- $1\kappa^2O,N:2\kappa^2N',O'$ -bis-[(furan-2-carbaldehyde azine- $\kappa^2N,O$ )silver(I)] bis-(hexafluorophosphate): an unusual complex containing two metal atoms and three ligands. Corrigendum**

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In the paper by Wang, Dong, Ma & Huang [*Acta Cryst.* (2005), E61, m2369–m2370], the correspondence author is incorrectly indicated. The correct correspondence author is given here, together with revised postal and e-mail addresses.

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## Key indicators

Single-crystal X-ray study  
 $T = 298$  K  
Mean  $\sigma(\text{C}-\text{C}) = 0.008$  Å  
 $R$  factor = 0.044  
 $wR$  factor = 0.111  
Data-to-parameter ratio = 12.8For details of how these key indicators were  
automatically derived from the article, see  
<http://journals.iucr.org/e>. **$\mu$ -(Furan-2-carbaldehyde azine)- $1\kappa^2\text{O},\text{N}:2\kappa^2\text{N}',\text{O}'$ -  
bis[(furan-2-carbaldehyde azine- $\kappa^2\text{N},\text{O}$ )silver(I)]  
bis(hexafluorophosphate): an unusual complex  
containing two metal atoms and three ligands**

The title complex,  $[\text{Ag}_2(\text{C}_{10}\text{H}_8\text{N}_2\text{O}_2)_3](\text{PF}_6)_2$ , contains two silver cations and three molecules of the new Schiff base ligand furan-2-carbaldehyde azine, accompanied by two charge-balancing  $\text{PF}_6^-$  anions. There is a centre of inversion at the mid-point of the N—N bond of the central ligand. The Ag atom adopts a 'see-saw' coordination, with two short Ag—N bonds and two long Ag—O bonds.

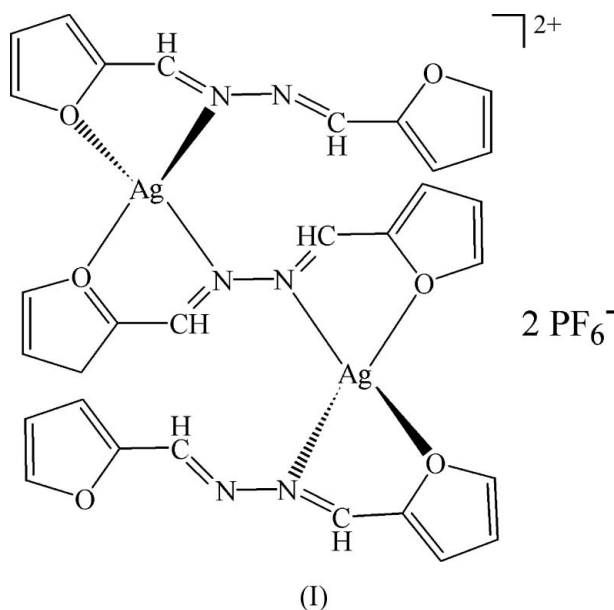
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## Comment

Combining metal ions with double Schiff base ligands may result in coordination polymers with novel network connectivities (Yu Bin *et al.*, 2005). Our interest in understanding the relationship between the metal coordination modes with such ligands and their extended structures led us to synthesize the title  $\text{Ag}^{\text{I}}$  complex, (I), and we report its structure here (Fig. 1).



Compound (I) contains a complex ion made up from the unusual combination of two metal ions and three furan-2-carbaldehyde azine (bdb) ligand molecules. There is an inversion centre at the mid-point of the N—N bond of the central ligand.

The Ag centre adopts a very distorted  $\text{AgN}_2\text{O}_2$  coordination geometry, arising from  $N,O$ -chelation by the Schiff base N atom and furanyl O atom from two bdb ligands (Table 1). Overall, the Ag coordination could be described as 'see-saw', if not simply irregular. A non-coordinated  $\text{PF}_6^-$  counter-ion occupying a general position completes the structure of (I).

Experimental

A methanol solution (8 ml) of AgPF<sub>6</sub> (25.3 mg, 0.1 mmol) was slowly diffused into a dichloromethane solution (8 ml) of 1,4-bis(furanyl)-2,3-diaza-1,3-butadiene (28.2 mg, 0.15 mmol). Colourless single crystals of (I) were obtained after the solution was allowed to stand at room temperature for three days.

Crystal data

[Ag<sub>2</sub>(C<sub>10</sub>H<sub>8</sub>N<sub>2</sub>O<sub>2</sub>)<sub>3</sub>](PF<sub>6</sub>)<sub>2</sub>  
*M<sub>r</sub>* = 1070.23  
 Triclinic, *P* $\bar{1}$   
*a* = 9.414 (2) Å  
*b* = 9.943 (2) Å  
*c* = 11.332 (3) Å  
 $\alpha$  = 94.829 (3)°  
 $\beta$  = 98.144 (3)°  
 $\gamma$  = 117.247 (2)°  
*V* = 920.0 (4) Å<sup>3</sup>  
*Z* = 1  
*D<sub>x</sub>* = 1.932 Mg m<sup>-3</sup>  
 Mo *K*α radiation  
 Cell parameters from 2033 reflections  
 $\theta$  = 2.3–26.7°  
 $\mu$  = 1.26 mm<sup>-1</sup>  
*T* = 298 (2) K  
 Slab, colourless  
 0.36 × 0.12 × 0.05 mm

Data collection

Bruker SMART CCD diffractometer  
 $\varphi$  and  $\omega$  scans  
 Absorption correction: multi-scan (SADABS; Bruker, 1997)  
*T<sub>min</sub>* = 0.659, *T<sub>max</sub>* = 0.940  
 4924 measured reflections  
 3346 independent reflections  
 2888 reflections with *I* > 2σ(*I*)  
*R<sub>int</sub>* = 0.023  
 $\theta_{max}$  = 25.5°  
*h* = -11 → 11  
*k* = -12 → 11  
*l* = -13 → 9

Refinement

Refinement on *F*<sup>2</sup>  
*R* [*F*<sup>2</sup> > 2σ(*F*<sup>2</sup>)] = 0.044  
*wR* (*F*<sup>2</sup>) = 0.111  
*S* = 1.05  
 3346 reflections  
 262 parameters  
 H-atom parameters constrained  
 $w = 1/[\sigma^2(F_o^2) + (0.0581P)^2 + 0.2485P]$   
 where  $P = (F_o^2 + 2F_c^2)/3$   
 $(\Delta/\sigma)_{max} < 0.001$   
 $\Delta\rho_{max} = 0.69 \text{ e \AA}^{-3}$   
 $\Delta\rho_{min} = -0.49 \text{ e \AA}^{-3}$

Table 1

Selected geometric parameters (Å, °).

Ag1–N2	2.194 (3)	Ag1–O1	2.764 (5)
Ag1–N3	2.243 (3)	Ag1–O2	2.700 (6)
N2–Ag1–N3	158.35 (12)	O1–Ag1–O2	68.68 (10)
N2–Ag1–O1	67.95 (12)	N2–Ag1–O2	117.87 (12)
N3–Ag1–O1	97.57 (12)	N3–Ag1–O2	67.59 (11)

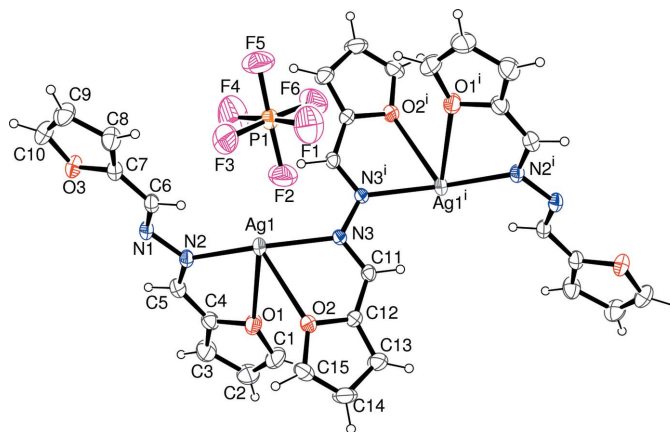


Figure 1

View of (I), showing 30% displacement ellipsoids. Symmetry code: (i): 1 - *x*, 1 - *y*, 1 - *z*.

H atoms bonded to C atoms were included in calculated positions and refined as riding [C–H = 0.93 Å; *U<sub>iso</sub>*(H) = 1.2*U<sub>eq</sub>*(C)].

Data collection: SMART (Bruker, 1997); cell refinement: SAINT (Bruker, 1997); data reduction: SAINT; program(s) used to solve structure: SHELXTL (Bruker, 2001); program(s) used to refine structure: SHELXTL; molecular graphics: SHELXTL; software used to prepare material for publication: SHELXTL.

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